Manufacturing Process with Various Determinants and Properties of Ordinary Portland Cement

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Abstract

The OPC is silicate cement normally utilized for concrete masonry units and for all uses where the concrete isn't dependent upon unique sulfate hazard or where the heat produced by the hydration of cement isn't objectionable. It has incredible resistance from breaking but has a smaller amount of resistance from chemical attacks. Keywords: OPC, Limestone, clinker etc.

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I. INTRODUCTION

It is characterized as the bonding material taking consistent and cement assets which makes it able to join the diverse development materials. OPC is accepted to be future item considering of it as different applications in the development business. They are appropriate to use in hostile environmental conditions. They can be dependably used in the development of marine structures, masonry mortars and plastering, hydraulic structures. Also, they are prevalently utilized in mass concreting works, for example, dykes, sewage pipes, dams, and so on.

II. MANUFACTURE OF OPC CEMENT

Cement is a hydraulic binder and is defined as a finely ground inorganic material which, when mixed with water, forms a paste which sets and hardens by means of hydration reactions and processes which, after hardening retains it's strength and stability even under water. OPC is one of a few kind of cement being produced all through the world. Magnesium (MgO) is available in little amounts as a polluting influence related with limestone. SO3 is added at the grinding stage to impede the setting time of the completed cement. At the point when cement raw materials containing the best possible extents of the basic oxides are ground to an appropriate fineness and afterward consumed to nascent combination in an oven, chemical mix happens, to a great extent

in the strong state bringing about an item suitably named clinker. Significantly there are 4 stages associated with the OPC cement,

1. Crushing and grinding of raw material

Firstly, Soft raw materials are squashed into reasonable size. It is completed for the most part in cylindrical ball holding the charge of steel balls.

2. Mixing or Blending

Now the crushed limestone is blended in with the clay paste in legitimate extent (75%=lime stone; clay=25%). In the dry process these mixes are stored in silos; slurry tanks are used in the wet process. Silos and slurry tanks can be used to put away these mix in dry and wet process respectively

3. Heating

The rotational kiln comprises of big cylinders 8 to 15 feet in diameter & height of 300-500 feet. Firstly the mix is preheated to 550C, where all the dampness content is evaporated and clay is broken into silica, aluminum oxide, iron oxide.

In the following zone the temperature is raised to 1500 degree Celsius where the oxides structure separate silicate, aluminates and ferrite. In the last step the item is chilled off to 200C. Kiln rotates at the rate of 1-2 revolution per minute. In rotary kiln, slurry is passed through different zones of temperature. This whole process in kiln usually covers 2 to 3 hours.

4. Grinding

The Cement Clinker is then air cooled. the cement clinkers and gypsum grinded into fine powders which are put away in the storehouses. At long last cement is filled in bags and afterward moved to the necessary site.

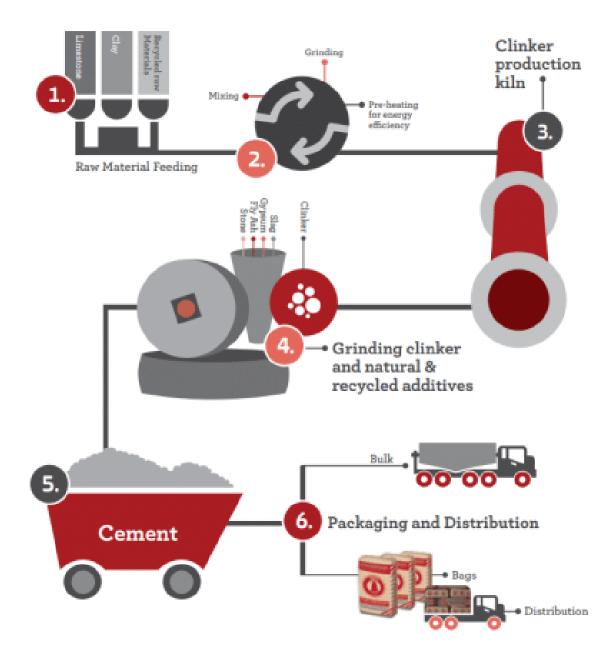


Fig 1: Manufacturing process of ordinary Portland cement

Constituent	OPC(% by weight)
Lime	65.64
Silica	22.20
Alumina	5.53
Iron oxide	3.36
Magnesia	2.08
Sulphu Trioxide	2.23
N ₂ O	0.06
Loss of Ignition	0.62
Lime saturation factor	0.95

TABLE 1VARIOUS DETERMINANTS IN OPC

TABLE 2PROPERTIES OF OPC CEMENT

Properties	Values
Specific Gravity	3.12
Normal Consistency	29%
Initial Setting time	65min
Final Setting time	275 min
Fineness	330 kg/m ²
Soundness	2.5mm
Bulk Density	830-1650 kg/m ³

III. CONCLUSION

The raw materials for Portland cement are basically made out of three segments: calcium oxide, silicon dioxide and alumina, which account for 60%, 20% and 10% of the total component separately. The calcium oxide originates from limestone, while silica and alumina originate from shale, dirt and bauxite. In addition, the majority of the raw materials contain iron oxide, magnesia, calcium sulfate, and so on.

REFERENCES

- [1] C. M. Hanson, "Concrete: the advanced industrial material of the 21st century," Metallurgical & Materials Transactions A, vol. 26, pp. 1321–1341, 1995.
- [2] M. Bediako, S. K. Y. Gawu, and A. A. Adjaottor, "Suitability of some Ghanaian mineral admixtures for masonry mortar formulation," Construction and Building Materials, vol. 29, pp. 667–671, 2012.
- [3] S. H. Kosmatka, B. Kerkhoff, and W. C. Panarese, Design and Control of Concrete Mixtures, Portland Cement Association, Skokie, Ill, USA, 14th edition, 2002.
- [4] M. S. Mamlouk and J. P. Zaniewski, Materials for Civil andConstruction Engineers, Prentice Hall, Upper Saddle River, NJ,USA, 2006.
- [5] T. Punmatharith, M. Rachakornkij, A. Imyim, and M.Wecharatana, "Co-processing of grinding sludge as alternative raw material in portland cement clinker production," Journal of Applied Sciences, vol. 10, no. 15, pp. 1525–1535, 2010.
- [6] D. N. Huntzinger and T. D. Eatmon, "A life-cycle assessment of Portland cement manufacturing: comparing the traditional process with alternative technologies," Journal of Cleaner Production, vol. 17, no. 7, pp. 668–675, 2009.
- [7] F. M. Lea, The Chemistry of Cement and Concrete, Arnold Publishers, London, UK, 3rd edition, 1970.
- [8] J. F. Young, S.Mindess, R. J. Gray, and A. Bentur, TheScience and Technology of Civil Engineering Materials, Prentice-Hall, Upper Saddle River, NJ, USA, 1998.
- [9] H. F. W. Taylor, Cement Chemistry, Thomas Telford, London, UK, 2nd edition, 1997.
- [10] S. H. Kosmatka and M. L. Wilson, Design and Control of Concrete Mixtures, Portland Cement Association, Stokie, Ill, USA, 2011.
- [11] V. Sata, C. Jaturapitakkul, and K. Kiattikomol, "Influence of pozzolan from various by-product materials on mechanical properties of high-strength concrete," Construction and BuildingMaterials, vol. 21, no. 7, pp. 1589–1598, 2007.
- [12] A. Neville, Neville on Concrete, ACI, Farmington Hills, Mich, USA, 2003.
- [13] R. Fernandez, F. Martirena, and K. L. Scrivener, "The origin of the pozzolanic activity of calcined clay minerals: a comparison between kaolinite, illite and montmorillonite," Cement and Concrete Research, vol. 41, no. 1, pp. 113–122, 2011.
- [14] K. Ganesan, K. Rajagopal, and K. Thangavel, "Evaluation of bagasse ash as supplementary cementitious material," Cement and Concrete Composites, vol. 29, no. 6, pp. 515–524, 2007.
- [15] S. Sinthaworn and P. Nimityongskul, "Quick monitoring of pozzolanic reactivity of waste ashes," Waste Management, vol.29, no. 5, pp. 1526–1531, 2009.

- [16] J. J. Brooks, M. A.M. Johari, and M.Mazloom, "Effect of admixtures on the setting times of highstrength concrete," Cement and Concrete Composites, vol. 22, no. 4, pp. 293–301, 2000.
- [17] X. Fu, Z.Wang,W. Tao et al., "Studies on blended cement with a large amount of fly ash," Cement and Concrete Research, vol. 32,no. 7, pp. 1153–1159, 2002.
- [18] E.-H. Kadri, S. Kenai, K. Ezziane, R. Siddique, and G. De Schutter, "Influence of metakaolin and silica fume on the heat of hydration and compressive strength development of mortar," Applied Clay Science, vol. 53, no. 4, pp. 704–708, 2011.
- [19] V. Indrawati and A. Manaf, "Mechanical strength of trass as supplementary cementing material," Journal of Physical Science, vol. 92, no. 2, pp. 51–59, 2008.